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60/428,180 21 November 2002 (21.11.2002) US(71) Applicant (for all designated States except US): **SEPRACOR INC.** [US/US]; 84 Waterford Drive, Marlborough, MA 01752 (US).

(72) Inventors; and

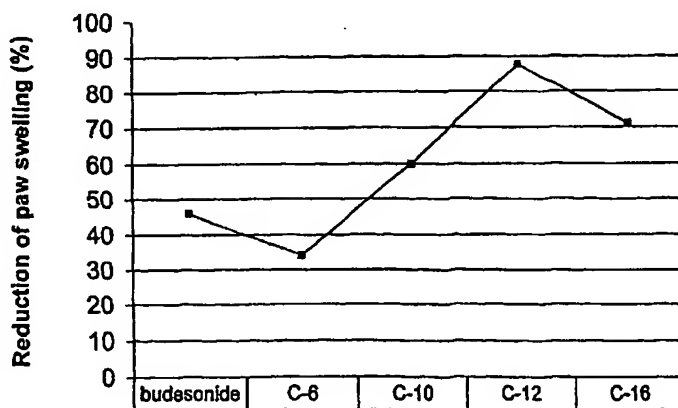
(75) Inventors/Applicants (for US only): **GROVER, Paul**[US/US]; 4 Warwick Road, Franklin, MA 02038 (US).  
**JONES, Steve** [US/US]; 10 Ragged Hill Road, Milford, MA 01757 (US). **SENANAYAKE, Chris, H.** [US/US]; 12 Isabel's Way, Brookfield, CT 06804 (US). **CURRIE, Mark, G.** [US/US]; 18 Hall Avenue, Sterling, MA 01564 (US). **FANG, Q., Kevin** [US/US]; 35 Atwood Street, Wellesley, MA 02181 (US).(74) Agent: **HANSEN, Philip E.**; HESLIN ROTHENBERG FARLEY & MESITI P.C., 5 Columbia Circle, Albany, NY 12203 (US).

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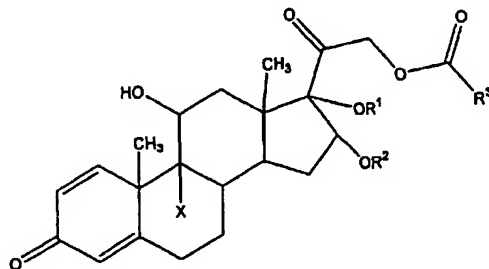
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(54) Title: CARBONATE AND CARBAMATE MODIFIED FORMS OF GLUCOCORTICOIDS

Effect of compounds on rat paw edema



(57) Abstract: Carbonates and carbamates of the formula and related steroid carbonates and carbamates are disclosed. The compounds are useful for treating rhinitis and asthma, particularly by inhalation, and for treating inflammation, particularly by local or topical administration.



(I)



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## CARBONATE AND CARBAMATE MODIFIED FORMS OF GLUCOCORTICOIDS

### Field of the Invention

[0001] The invention relates to antiasthmatic carbonate and carbamate derivatives of glucocorticoids.

### Background of the Invention

[0002] Glucocorticoids, in topical, oral and inhaled formulations, are useful for their anti-inflammatory and immunosuppressive activities. Notwithstanding the sophistication of many formulations, many glucocorticoids exhibit significant side-effects that prevent realization of their maximum pharmacologic value. These side-effects stem, in part, from the difficulty of effectively delivering the glucocorticoid drug to a target tissue without increasing systemic concentrations of the drug.

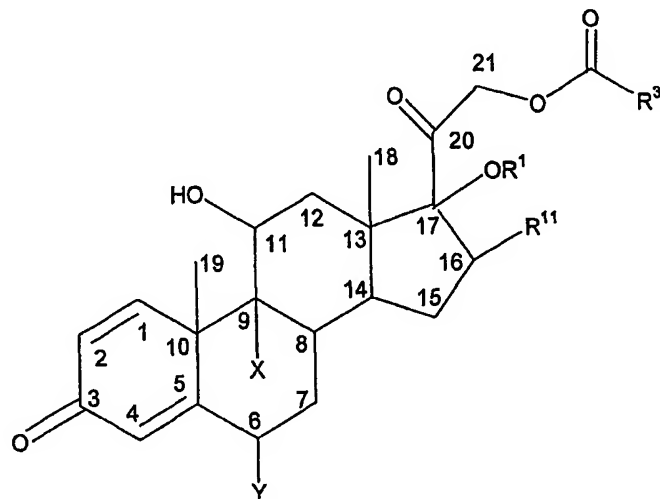
[0003] Inhaled glucocorticoids are an effective therapy for the control of asthma, and improvement with steroids is one of the hallmarks of asthma [Barnes, PJ (1998) in *Asthma: Basic Mechanisms and Clinical Management* (3<sup>rd</sup> ed)]. The inhaled glucocorticoids work to reduce the inflammation in either lungs, *e.g.* for asthma, or nose, *e.g.* for nasal allergies. Inhaled glucocorticoids are most often administered using a metered dose inhaler (MDI). In the best of circumstances, in controlled clinical settings, only around 30% of the administered dose gets into the lungs. In the general patient population probably only 10% or so of the dose gets into the lungs due to improper use of the inhaler. The rest of the administered drug is deposited in the throat and upper airways, or is swallowed. The drug that is deposited in the throat is responsible for some side effects seen with inhaled glucocorticoids (cough, oropharyngeal candidiasis and dysphonia). For early generation inhaled glucocorticoids, the swallowed drug leads to the same side effects seen

with oral glucocorticoids. In light of the tremendous efficacy of inhaled glucocorticoids in asthma, much effort has gone into reducing the side effects from their use. Although newer glucocorticoids (e.g. budesonide, ciclesonide, triamcinolone and fluticasone) exhibit reduced systemic side effects from swallowed drug - being either poorly absorbed in the gut or subject to extensive inactivation in the liver - they nonetheless display systemic side effects as a result of absorption from the lung into the systemic circulation. The side effects include decreased bone density (Israel, E *et al.*, (2001), *New England Journal of Medicine* 345:941-947 and Wong, CA *et al.*, (2000) *Lancet* 355:1399-1403), which has been correlated with increased risk of fracture. Thus the need still exists for inhaled glucocorticoids with reduced systemic effects.

[0004] Several approaches have been suggested to reduce systemic effects. One such approach takes advantage of inactive prodrugs that are activated in the lung tissues. For example, Dietzel *et al.* [*Prog. Respir Res.* 31, 91-93 (2001)] have described an isopropyl group esterified at the 21 position of the glucocorticoid core structure. Another approach that has been suggested is the formulation of a glucocorticoid as a liposome. Axelsson *et al.* in a series of US patents (4,693,999; 5,614,514 and 5,888,995) describe selected glucocorticoids modified for formulation into liposomes by esterification at the 21 position with saturated and mono-unsaturated fatty acids with chain lengths up to 20 carbons.

#### Summary of the Invention

It has now been found that carbonates and carbamates of Formula I:



(I)

provide unexpectedly greater potency for an extended time than do either the parent alcohols or the shorter chain carbonates and carbamates. In the compounds of formula I according to the invention:

$R^1$  and  $R^2$ , independently for each occurrence, represent a hydrogen, lower alkyl or lower acyl, or taken together  $R^1$  and  $R^2$  form a substituted or unsubstituted ketal;

$R^3$  is  $-OR^4$  or  $-NR^5R^6$ ;

$R^4$  is chosen from  $C_7$  to  $C_{24}$  hydrocarbon,  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-COOH$  and  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-NR^9R^{10}$ ;

$R^5$  is hydrogen or  $C_7$  to  $C_{24}$  hydrocarbon;

$R^6$  is chosen from  $C_7$  to  $C_{24}$  hydrocarbon and  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-COOH$ ;

$R^9$  is hydrogen or  $C_1$  to  $C_{17}$  hydrocarbon;

$R^{10}$  is hydrogen or  $C_1$  to  $C_{17}$  hydrocarbon;

$R^{11}$  is methyl or  $-OR^2$ ; and

X and Y are independently hydrogen or halogen.

[0005] In another aspect the invention relates to methods for treating rhinitis, asthma and inflammatory diseases and conditions comprising administering the compounds of formula I.

[0006] In another aspect, the invention relates to pharmaceutical formulations for inhalation comprising the compounds of formula I, a pharmaceutically acceptable fluid for suspension or solution, and, for metered dose inhalers, additionally comprising a propellant.

#### Brief Description of the Drawings

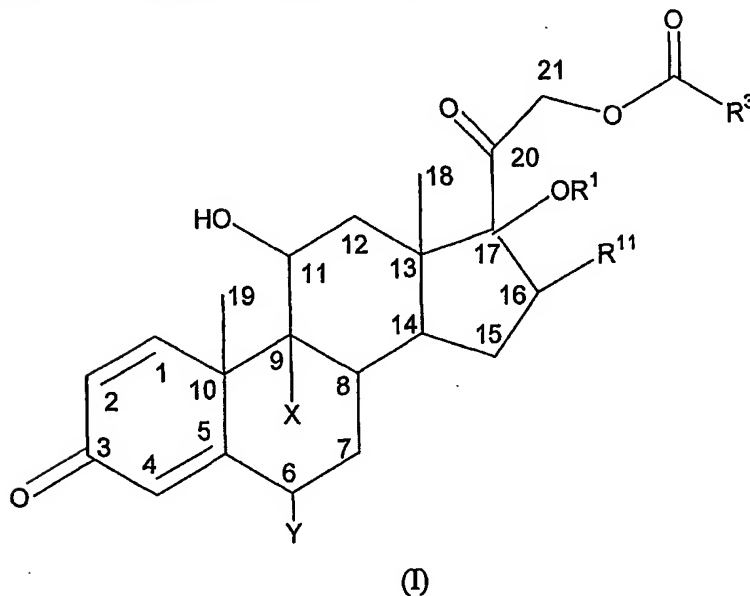
[0007] Figure 1 is a graph of the percent reduction in rat paw edema as a function of the number of carbons in a series of carbamate esters of budesonide.

[0008] Figure 2 is a graph of the percent reduction in rat paw edema as a function of the dose of budesonide at four time intervals.

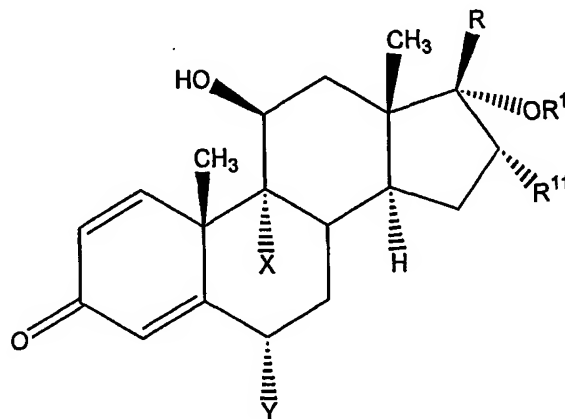
[0009] Figure 3 is a graph of the percent reduction in rat paw edema as a function of the dose of budesonide dodecylcarbonate at four time intervals.

#### Detailed Description of the Invention

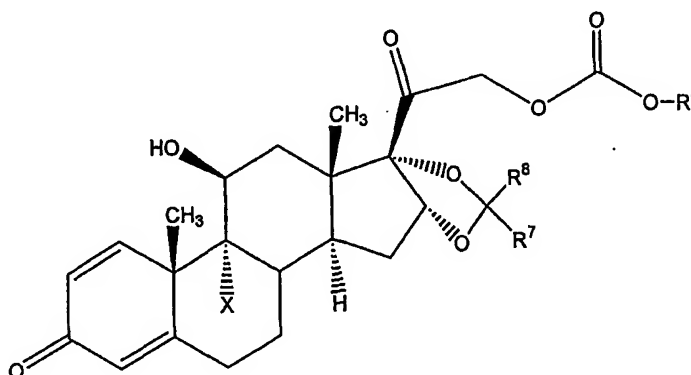
[0010] The invention relates to compounds of Formula I:



in which the substituents are as defined above. In preferred embodiments the steroid has the absolute stereochemistry shown:



**[0011]** Examples of steroids having the foregoing structure include budesonide, ciclesonide, fluticasone and triamcinolone. The most preferred embodiment comprises compounds of formula:

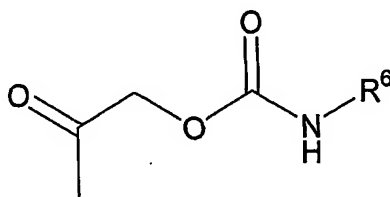


**[0012]** wherein R<sup>7</sup> is hydrogen or lower alkyl; and R<sup>8</sup> is lower alkyl. In particularly preferred embodiments, R<sup>4</sup> is C<sub>11</sub> to C<sub>14</sub> alkyl, C<sub>12</sub> to C<sub>24</sub> alkyl, C<sub>12</sub> to C<sub>20</sub> alkyl, C<sub>7</sub> to C<sub>24</sub> alkyl, C<sub>8</sub> to C<sub>24</sub> alkyl, C<sub>9</sub> to C<sub>24</sub> alkyl, C<sub>10</sub> to C<sub>24</sub> alkyl, C<sub>11</sub> to C<sub>24</sub> alkyl, C<sub>8</sub> to C<sub>18</sub> alkyl, C<sub>10</sub> to C<sub>16</sub> alkyl or C<sub>8</sub> to C<sub>20</sub> alkyl. In preferred embodiments the steroid is budesonide, ciclesonide or triamcinolone. Budesonide dodecyl carbonate is most preferred.



[0013] In embodiments in which  $R^3$  is  $-OR^4$  and  $R^4$  is  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-NR^9R^{10}$ , it is preferred that the total number of carbons in  $R^3$  be eight to twenty-four. Similarly, in embodiments in which  $R^3$  is  $-NR^5R^6$ , it is preferred that the sum of the number of carbons in  $R^5$  plus the number of carbons in  $R^6$  be seven to twenty-four. The underlying guideline is that the total number of carbons in the residue  $R^3$  is optimally seven to twenty-four, but an amino function could be interposed at any point that results in an  $R^3$  residue that is chemically stable in combination with the adjacent  $O(C=O)$  residue.

[0014] In another particularly preferred embodiment,  $R^3$  is



and  $R^6$  is  $C_{11}$  to  $C_{14}$  alkyl,  $C_{12}$  to  $C_{24}$  alkyl,  $C_{12}$  to  $C_{20}$  alkyl,  $C_7$  to  $C_{24}$  alkyl,  $C_8$  to  $C_{24}$  alkyl,  $C_9$  to  $C_{24}$  alkyl,  $C_{10}$  to  $C_{24}$  alkyl,  $C_{11}$  to  $C_{24}$  alkyl,  $C_8$  to  $C_{18}$  alkyl,  $C_{10}$  to  $C_{16}$  alkyl or  $C_8$  to  $C_{20}$  alkyl.

[0015] Alkyl is intended to include linear, branched, or cyclic hydrocarbon structures and combinations thereof. Preferred alkyl groups are those of  $C_7$  to  $C_{24}$ . Cycloalkyl is a subset of alkyl and includes cyclic hydrocarbon groups, in this case preferably from 6 to 8 carbon atoms. Lower acyl is acyl of one to six carbons, e.g. acetyl, propionyl, isopropanoyl, butanoyl, sec-butanoyl, valeroyl, and hexanoyl.

[0016]  $C_7$  to  $C_{24}$  Hydrocarbon includes alkyl, cycloalkyl, alkenyl, alkynyl, aryl and combinations thereof. Examples include phenethyl, cyclohexylmethyl, camphoryl and naphthylethyl.

[0017] The compounds described herein contain one or more asymmetric centers and may thus give rise to enantiomers, diastereomers, and other stereoisomeric forms that may be defined, in terms of absolute stereochemistry, as (R)- or (S)-. The present invention is meant to include all such possible isomers, as well as, their racemic and optically pure forms. Optically active isomers may be prepared using chiral synthons or chiral reagents, or resolved using conventional techniques. When the compounds described herein contain olefinic double bonds or other centers of geometric asymmetry, and unless specified otherwise, it is intended that the compounds include both E and Z geometric isomers. Likewise, all tautomeric forms are also intended to be included. The graphic representations of racemic, ambiscalemic and scalemic or enantiomerically pure compounds used herein are taken from Maehr J. Chem. Ed. 62, 114-120 (1985): solid and broken wedges are used to denote the absolute configuration of a chiral element; wavy lines indicate disavowal of any stereochemical implication which the bond it represents could generate; solid and broken bold lines are geometric descriptors indicating the relative configuration shown but denoting racemic character; and wedge outlines and dotted or broken lines denote enantiomerically pure compounds of indeterminate absolute configuration.

[0018] The abbreviations Me, Et, Ph, Tf, Ts and Ms represent methyl, ethyl, phenyl, trifluoromethanesulfonyl, toluensulfonyl and methanesulfonyl respectively. A comprehensive list of abbreviations utilized by organic chemists (i.e. persons of ordinary skill in the art) appears in the first issue of each volume of the Journal of Organic Chemistry. The list, which is typically presented in a table entitled "Standard List of Abbreviations", provides the basis for the abbreviations not otherwise defined herein.

[0019] The term "methods of treating" when used in connection with the present invention means amelioration, prevention or relief from the symptoms and/or effects associated with asthma and rhinitis. The person of ordinary skill in the medical art recognizes that "prevention" of the symptoms and/or effects

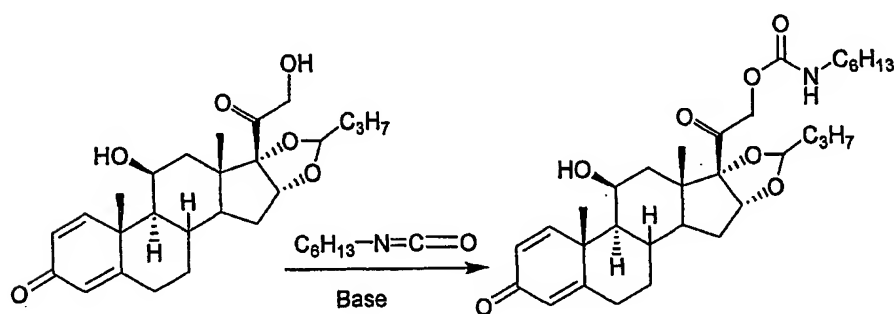
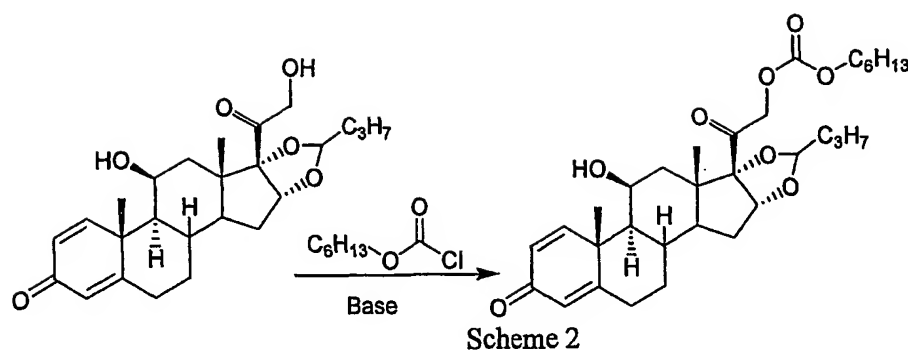
associated with asthma and rhinitis is not an absolute term. In the medical art it is understood to refer to the prophylactic administration of a drug to substantially diminish the likelihood or seriousness of the condition.

[0020] The compounds of the invention are useful for treating asthma and rhinitis. They are also useful for intra-articular injection for alleviating the joint pain, swelling and stiffness associated with rheumatoid arthritis and osteoarthritis with an inflammatory component; also for bursitis, epicondylitis and tenosynovitis. They may be used topically, transdermally and intradermally (intra-lesional) in lichen simplex chronicus, granuloma annulare, lichen planus, keloids, alopecia areata, discoid lupus erythematosus, localised neurodermatitis, cystic acne, granuloma annulare, nummular and dyshydrotic eczema, and hypertrophic scars (keloids). The treatment of macular degeneration with compounds of the invention is analogous to that described in Billson, US patent 5,770,589, which is incorporated herein by reference.

[0021] In general, the compounds of the present invention may be prepared by the methods illustrated in the general reaction schemes as, for example, described below, or by modifications thereof, using readily available starting materials, reagents and conventional synthesis procedures. In these reactions, it is also possible to make use of variants that are in themselves known, but are not mentioned here.

[0022] Exemplary syntheses of a budesonide carbonate and a carbamate are shown in Schemes 1 and 2. One skilled in the art will recognize that the syntheses can be adapted to prepare a variety of carbonate or carbamate modified budesonide, ciclesonide, fluticasone or triamcinolone analogs.

Scheme 1.



#### Synthesis of budesonide Dodecylcarbonate. (Example 2)

[0023] To the solution of budesonide (750 mg, 1.74 mmol) in DCM (7.5 mL) was added dodecyl chloroformate (513  $\mu$ L, 1.617 mmol) and  $\text{Et}_3\text{N}$  (533  $\mu$ L, 3.825 mmol) at room temperature. The reaction mixture was stirred at room temperature for 7 hours. During this 7 hours more dodecyl chloroformate (510  $\mu$ L, 1.616 mmol) and  $\text{Et}_3\text{N}$  (440  $\mu$ L, 3.18 mmol) were added. The reaction was followed by HPLC. The reaction mixture was poured into water (20 mL) and DCM (10 mL); the aqueous phase was extracted with DCM (10 mL). The combined organic phases were washed with water (10 mL) and brine (10 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated in *vacuo* to provide crude budesonide dodecylcarbonate. The product was purified by chromatography, eluted with Hexane: AcOEt = 9:1 to 3:1 to provide 983 mg of 1:1 mixture of epimers (originated from budesonide) (96.78 area % purity on HPLC).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.80-2.25 (m, 45H), 2.36 (d, 1H, 13.4Hz), 2.58 (t, 1H,

13.2Hz), 4.19 (t, 2H, 6.7 Hz), 4.52 (bs, 1H), 4.6-5.2 (m, 5H), 6.03 (s, 1H), 6.30 (d, 1H, 10.1 Hz), 7.29 (d, 1H, 10.1 Hz).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.21, 14.38, 17.22, 17.35, 17.50, 17.78, 21.35, 22.94, 25.86, 28.83, 29.45, 29.60, 29.74, 29.80, 29.88, 30.57, 31.24, 32.16, 33.18, 33.68, 34.26, 35.26, 37.36, 41.17, 41.41, 44.24, 46.18, 47.61, 50.01, 53.14, 55.41, 55.52, 69.12, 69.18, 70.11, 70.22, 82.32, 83.57, 97.76, 98.62, 104.88, 108.63, 122.83, 128.24, 156.14, 169.81, 169.92, 186.74, 202.16 and 203.46. Mass spectrum ( $m/e$ ) 643 ( $M^+$ ).

#### Synthesis of budesonide hexadecylcarbonate. (Example 7)

[0024] To the solution of budesonide (431 mg, 1.0 mmol) in DCM (4.5 mL) was added hexadecyl chloroformate (655  $\mu\text{L}$ , 2.0 mmol) and  $\text{Et}_3\text{N}$  (512  $\mu\text{L}$ , 3.7 mmol) at room temperature. After reaction mixture was stirred at room temperature over night, it was poured into water (20 mL) and DCM (10mL); the aqueous phase was extracted with DCM (10mL). The combined organic phases were washed with water (10mL) and brine (10mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated in *vacuo* to provide crude budesonide hexadecylcarbonate. The product was purified by chromatography, eluted with Hexane: AcOEt = 9:1 to 3:1 to provide 432 mg of 1:1 mixture of epimers (originated from budesonide) (99.38 area % purity on HPLC).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.80-2.40 (m, 53H), 2.56 (dt, 1H, 13.1 and 4.6 Hz), 4.30 (t, 2H, 6.7 Hz), 4.5-5.2 (m, 7H), 6.00 (s, 1H), 6.26 (d, 1H, 10.1 Hz), 7.3 (d, 1H, 10.1 Hz).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.16, 14.18, 14.34, 17.10, 17.27, 17.39, 17.71, 21.21, 22.88, 25.81, 28.79, 29.41, 29.56, 29.70, 29.76, 29.87, 30.54, 31.19, 32.11, 33.08, 33.59, 34.23, 35.19, 37.28, 40.68, 40.98, 44.37, 46.08, 47.52, 50.00, 53.07, 55.39, 55.49, 68.98, 69.73, 69.83, 69.95, 70.04, 82.10, 83.38, 97.71, 98.58, 104.70, 108.51, 122.57, 127.91, 155.08, 156.85, 170.48, 170.59, 186.84, 186.88, 202.10 and 203.38. Mass spectrum ( $m/e$ ) 699 ( $M^+$ ).

#### Synthesis of budesonide decylcarbonate. (Example 8)

[0025] To the solution of budesonide (431 mg, 1.0 mmol) in DCM (4.5 mL) was added decyl chloroformate (460  $\mu\text{L}$ , 2.0 mmol) and  $\text{Et}_3\text{N}$  (512  $\mu\text{L}$ , 3.7

mmol) at room temperature. After the reaction mixture was stirred at room temperature over night, it was poured into water (20 mL) and DCM (10mL); the aqueous phase was extracted with DCM (10mL). The combined organic phases were washed with water (10mL) and brine (10mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in *vacuo* to provide crude budesonide decylcarbonate. The product was purified by chromatography, eluted with Hexane: AcOEt = 9:1 to 3:1 to provide 348 mg of 1:1 mixture of epimers (originated from budesonide) (99.22 area % purity on HPLC). <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.85-2.24 (m, 42H), 2.36 (dd, 1H, 13.4 and 2.9 Hz), 2.59 (dt, 1H, 13.5 and 4.5 Hz), 4.18 (t, 1H, 6.7 Hz), 4.51 (s, 1H), 4.6-5.2 (m, 5H), 6.03 (s, 1H), 6.30 (d, 1H, 10.1 Hz), 7.49 (d, 1H, 10.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.14, 14.17, 14.31, 17.06, 17.24, 17.36, 17.68, 21.19, 22.83, 25.78, 28.76, 29.37, 29.40, 29.66, 30.52, 31.16, 32.04, 33.05, 33.56, 34.22, 35.16, 37.25, 40.58, 40.89, 44.38, 46.06, 47.49, 49.98, 53.05, 55.38, 55.48, 68.90, 68.94, 69.65, 69.75, 69.92, 70.01, 82.05, 83.34, 97.69, 98.56, 104.65, 108.47, 122.51, 127.84, 155.04, 156.96, 170.61, 170.71, 186.84, 186.89, 202.08 and 203.37. Mass spectrum (m/e) 615 (M<sup>+</sup>).

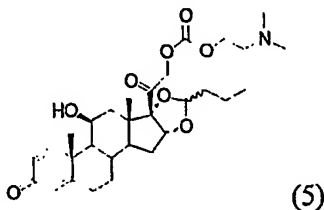
#### Synthesis of budesonide butylcarbamate. (Example 6)

[0026] To the solution of budesonide (500 mg, 1.16 mmol) in DCM (5.0 mL) was added butyl isocyanate (144 μL, 1.28 mmol) and DMAP (312 mg, 2.55 mmol) at room temperature. The reaction mixture was stirred at room temperature for 24 hours. During this 24 hours more butyl isocyanate (72 μL, 0.64 mmol) and DMAP (156 mg, 1.27 mmol). The reaction was followed by HPLC. The reaction mixture was poured into water (20 mL) and DCM (10 mL); the aqueous phase was extracted with DCM (10 mL). The combined organic phases were washed with water (10 mL) and brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in *vacuo* to provide crude budesonide butylcarbamate. The product was purified by chromatography, eluted with Hexane: AcOEt = 9:1 to 3:1 to provide 560 mg of 1:1 mixture of epimers (originated from budesonide) (99.22 area % purity on HPLC). <sup>1</sup>H NMR

(CDCl<sub>3</sub>)  $\delta$  0.82-2.7 (m, 30H), 3.06-3.20 (m, 3H), 4.45 (m, 1H), 4.58 (t, 1H, 4.5 Hz), 4.75-5.36 (m, 5H), 5.99 (s, 1H), 6.24 (d, 1H, 10.1 Hz), 7.29 (d, 1H, 10.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  13.98, 14.08, 14.22, 17.09, 17.30, 17.37, 17.73, 20.09, 20.28, 21.26, 30.56, 31.22, 32.11, 32.59, 33.13, 33.65, 34.27, 35.23, 37.34, 40.30, 40.73, 40.99, 41.17, 44.37, 46.09, 47.49, 50.01, 53.15, 55.43, 55.52, 67.82, 69.82, 69.93, 82.11, 83.37, 97.84, 98.75, 104.66, 108.43, 122.61, 127.97, 155.94, 156.03, 156.79, 170.45, 170.56, 186.95, 203.89 and 205.21.

Synthesis of budesonide 2-dimethylaminoethyl carbonate. (Example 20)

[0027] To the solution of budesonide (600 mg, 1.394 mmol) in DCM (5.0 mL) was added CDI (249 mg, 1.53 mmol) at RT. After 3 hr stirring, N, N-dimethylethanolamine (308  $\mu$ L, 3.07 mmol) was added at room temperature. After the reaction mixture was stirred at room temperature for 3.5 hours, it was poured into water (20 mL) and DCM (10 mL); the aqueous phase was extracted with DCM (10 mL). The combined organic phases were washed with water (10 mL) and brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in *vacuo* to provide crude budesonide 2-dimethylaminoethyl carbonate. The product was purified by chromatography, eluted with AcOEt, then AcOEt: MeOH = 98:2 to 95:5 to provide 307 mg of 1:1 mixture of epimers (originated from budesonide) (98.49 area % purity on HPLC). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.80-2.55 (m, 29H), 3.69 (bs, 1H), 4.17 (t, 2H, 5.8 Hz), 4.42 (bs, 1H), 4.4-4.95 (m, 3H), 5.05 (dd, 1H, 12.5 and 7.2 Hz), 5.92 (s, 1H), 6.28 (d, 1H, 10.1 Hz), 7.29 (d, 1H, 10.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  14.16, 14.21, 17.08, 17.20, 17.37, 17.69, 21.22, 30.52, 31.18, 32.10, 33.06, 33.60, 34.28, 35.16, 37.24, 40.65, 40.94, 44.46, 45.86, 46.23, 47.64, 49.99, 53.11, 55.36, 55.45, 57.56, 66.27, 66.30, 69.34, 69.45, 70.14, 70.18, 82.14, 83.41, 97.76, 98.62, 104.64, 108.42, 122.48, 127.73, 154.92, 157.26, 157.37, 170.75, 170.89, 187.00, 187.08, 201.97 and 203.29.



The following compounds were synthesized as described above:

Example	Name	mol. Wt.
1	Budesonide isobutylcarbonate	530.65
2	Budesonide dodecylcarbonate	642.87
3	Budesonide hexylcarbonate	558.71
4	Budesonide dibutylcarbamate	585.77
5	Dexamethasone dodecylcarbonate	604.8
6	Budesonide butylcarbamate	529.67
7	Budesonide hexadecylcarbonate	698.98
8	Budesonide decylcarbonate	614.81
9	Budesonide hexylcarbonate	558.71
10	Budesonide dodecylcarbonate	642.86
11	Budesonide nonylcarbonate	600.78
12	Budesonide octylcarbonate	586.76
13	Budesonide undecylcarbamate	627.85
14	Budesonide heptylcarbonate	572.73
15	Budesonide 11-dimethylaminoundecylcarbonate	671.9
16	Budesonide phytolcarbonate	753.07
17	Budesonide farnesolcarbonate	678.90
18	Budesonide geraniolcarbonate	610.78
19	Budesonide nerolcarbonate	610.78

[0028] For administration, the drug is suitably inhaled from a nebulizer, from a pressurized metered dose inhaler or as a dry powder from a dry powder inhaler (e.g. sold as TURBUHALER®) or from a dry powder inhaler utilizing gelatin, plastic or other capsules, cartridges or blister packs.

[0029] A diluent or carrier, generally non-toxic and chemically inert to the medicament, e.g. lactose, dextran, mannitol or glucose or any additives that will give the medicament a desired taste, can be added to the powdered medicament.

[0030] Formulations and devices for nebulizers, metered dose inhalers and dry powder inhalers are well known to those skilled in the art. In formulations where the active ingredient is in a suspension it is important that the particles are below 20  $\mu\text{m}$  in size and preferably below 5  $\mu\text{m}$  in size. This may be achieved by micronization, crystallization, spray drying or other known techniques.



[0031] The solvent or suspension agent utilized for nebulization may be any pharmacologically suitable fluid such as water, aqueous saline, alcohols or glycols, e.g., ethanol, isopropylalcohol, glycerol, propylene glycol, polyethylene glycol, etc. or mixtures thereof. Saline solutions utilize salts which display little or no pharmacological activity after administration. Both inorganic salts, such as alkali metal or ammonium halogen salts e.g. sodium chloride, potassium chloride or organic salts, such as potassium, sodium and ammonium salts of organic acids, e.g., ascorbic acid, citric acid, acetic acid, tartaric acid, etc. may be used for this purpose.

[0032] Other excipients and additives may be added to the formulation. The active ingredient may be stabilized by the addition of an inorganic acid, e.g., hydrochloric acid, nitric acid, sulphuric acid and/or phosphoric acid; an organic acid, e.g., ascorbic acid, citric acid, acetic acid, and tartaric acid etc.; a complexing agent such as EDTA or citric acid and salts thereof; or an antioxidant such as vitamin E or ascorbic acid. These may be used alone or together to stabilize the active ingredient. Preservatives can also be added such as benzalkonium chloride or benzoic acid and salts thereof. Surfactant may be added particularly to improve the physical stability of suspensions. These include lecithins, disodium dioctylsulphosuccinate, oleic acid and sorbitan esters.

[0033] The active ingredient may also be suspended or dissolved in a liquified propellant, sealed in a container with a metering valve and fitted into an actuator. Such metered dose inhalers are well known in the art. The metering valve may meter 10 to 500  $\mu\text{L}$  and preferably 25 to 150  $\mu\text{L}$ .

[0034] The propellants used may be halocarbons, hydrocarbons or other liquified gasses. The most frequently used are trichlorofluoromethane (propellant 11), dichlorofluoromethane (propellant 12), dichlorotetrafluoroethane (propellant 114), tetrafluoroethane (HFA-134a), 1,1-difluoroethane (HFA-152a), difluoromethane (HFA-32), pentafluoroethane (HFA-125),

heptafluoropropane (HFA-227ea), perfluoropropane, perfluorobutane, perfluoropentane, butane, isobutane, and pentane. In particular, tetrafluoroethane (HFA-134a) and heptafluoropropane (HFA-227ea) and mixtures thereof are used.

[0035] As well as propellant, formulations may contain other excipients. Surfactant may be added particularly to improve the physical stability of suspensions and valve performance. These include lecithins, disodium dioctylsulphosuccinate, oleic acid and sorbitan esters. Cosolvents may also be added to improve solubility of surfactant in propellant or modify the pharmacological performance. These include alcohols and glycols, e.g., ethanol, isopropylalcohol, glycerol, propylene glycol, polyethylene glycol, etc., or mixtures thereof. Further excipients may be added to improve performance or taste, e.g., fatty acids and salts thereof such as magnesium stearate, menthol oil etc.

[0036] Dry powder inhalers include devices which meter drug from a chamber within the device or those that deliver pre-metered doses utilizing gelatin, plastic or other capsules, cartridges, or blister packs and/or strips.

[0037] For topical application, there are employed as non-sprayable forms, viscous to semi-solid or solid forms comprising a carrier compatible with topical application and having a dynamic viscosity preferably greater than water. Suitable formulations include but are not limited to solutions, suspensions, emulsions, creams, ointments, powders, liniments, salves, aerosols, etc., which are, if desired, sterilized or mixed with auxiliary agents, e.g., preservatives, stabilizers, wetting agents, buffers or salts for influencing osmotic pressure, etc. For topical application, also suitable are sprayable aerosol preparations wherein the active ingredient, preferably in combination with a solid or liquid inert carrier material, is packaged in a squeeze bottle or in admixture with a pressurized volatile, normally gaseous propellant, e.g., a freon.

[0038] The topical pharmaceutical carrier may include any substance capable of dispersing and maintaining contact between the active ingredients and the skin. The vehicle may be glycerin, alcohol or water based. Examples of such vehicles include aloe vera, which is a gel base, together with ethanol, isopropyl alcohol, water, propylene glycol and a non-ionic surfactant such as laureth-4. Other water-based alcohol/ glycerin vehicles and carriers are within the scope of the present invention. A typical water-based lotion will contain from 45 to 50 parts of glycerin, one to three parts Tween 80™, from 45 to 50 parts of water and from 1 to 50 parts of the compound of the invention.

[0039] Also included in the scope of the invention are ointments, emulsions or dispersions in which water, if present, is a minor constituent. Typical ointment formulation comprises from 90 to 98 parts of a mixture of petrolatum, mineral oil, mineral wax and wool wax alcohol, from 0.5 to 3 parts of a mixture of polyoxyethylene and sorbitan monooleate (Tween 80™), from 1 to 5 parts of water, and from 1 to 50 parts of the compound of the invention. Another suitable non-aqueous ointment can be prepared from 95 parts of liquid petrolatum USP, 5 parts polyethylene and from 1 to 50 parts of the compound of the invention. The resulting ointment spreads easily and has an even consistency over wide temperature extremes. It is, in addition, non-irritating and non-sensitizing.

[0040] Formulations of the compounds of the invention may also be prepared containing from 0 to 25% by weight of urea. In general, in such urea containing ointments, the water content will vary from 5 to 50% by weight of the composition. Any suitable ointment carrier may be used such as lanolin, ethylene glycol polymers and the like. In the case of formulations containing urea, it is known in the art that borate salts may often be added to stabilize the pharmaceutical composition (see U.S. patent 2,917,433, the disclosure of which is incorporated herein by reference).

[0041] Water based compositions may also be employed, in which case the compound of the invention will commonly be in solution, and the aqueous solution may, if desired, be thickened with a suitable gel to provide a less mobile composition. Such compositions are well known in the art

[0042] Compounds as described above were tested in the following assay for biological activity. The WI-38 human lung fibroblast line was obtained from the ATCC (catalog number 75-CCL) and maintained in Basal Medium Eagle with Earle's salts (GibcoBRL product number 21010-046) supplemented with 2 mM glutamine and 10% fetal calf serum at 37°C in a 7% CO<sub>2</sub> (balance air), humidified atmosphere. One week before experiments were done, the WI-38 cells were seeded into 48-well tissue culture dishes and maintained in media containing 10% fetal calf serum. The cells were used when confluent. The day before the experiment the cells were fed fresh media containing 10% fetal calf serum (0.25 mL per well). One the day of the experiment the media was removed from the cells and 0.25 mL of media containing 5% fetal calf serum added.

[0043] The rat alveolar macrophage cell line RAW 264.7 was obtained from the ATCC (catalog number 71-TIB) and maintained in Dulbecco's Modified Eagle Medium (GibcoBRL product number 11960-044) supplemented with 2 mM glutamine, 1 mM sodium pyruvate and 10% fetal calf serum at 37°C in a 10% CO<sub>2</sub> (balance air), humidified atmosphere. One week before experiments were done, the WI-38 cells were seeded into 48-well tissue culture dishes and maintained in media containing 10% fetal calf serum. The cells were used when confluent. The day before the experiment the cells were fed fresh media containing 10% fetal calf serum (0.25 mL per well). One the day of the experiment the media was removed from the cells and 0.25 mL of media containing 5% fetal calf serum added.

[0044] To determine the IC<sub>50</sub> values for the compounds, 1 to 1000 dilutions were made of the 5 mM stock solutions in DMSO to give 5 uM solutions.

These solutions were serially diluted 1:2 in DMSO to give a series of 12 dilutions ranging from 5  $\mu$ M to 2.4 nM. 0.0025 mL aliquots of the 12 dilutions were added to wells of the WI-38 cells to give final compound concentrations ranging from 50 nM to 0.024 nM. The cells were stimulated by addition of 0.001 mL of 0.025  $\mu$ g/mL recombinant human Interleukin-1 $\beta$  (IL-1 $\beta$  - Calbiochem catalog number 407615) in 0.1% bovine serum albumin in phosphate buffered saline. The cells were incubated for 24 hours and the supernatants harvested. The level of PGE<sub>2</sub> in the supernatants was assayed using a commercial Enzyme Immuno Assay (EIA) kit (Cayman Chemical catalog number 514010) after diluting 1:10 in EIA buffer according to the manufacturer's directions. The data from these experiments was fit to a 4 parameter logistic function using the IC<sub>50</sub> routine in the Grafit 4 program (Erithecus software). IC<sub>50</sub> values determined in this manner were: budesonide 0.20 nM; budesonide isobutylcarbonate 0.12 nM; budesonide dodecylcarbonate 0.53 nM; budesonide hexylcarbonate 0.14 nM.

[0045] The compounds of the invention were also tested *in vivo* in a rat paw edema model [Hirschelmann, R. and Bekemeier, H., Int J Tissue React 6, 471-475 (1984)], which persons of skill in the art accept as predictive of efficacy in treating asthma and rhinitis in humans.

[0046] Rat Paw Edema Protocol: Male Sprague Dawley rats Rj: SD (IOPS Han) (CEJ, France) weighing between 140 and 160 grams were used in the studies. Animals were housed in a temperature (19.5 – 24.5 °C), relative humidity (40 – 70%) and 12-hour light/dark cycle (light 6:00 a.m. to 6:00 p.m.)-controlled room, with *ad libitum* access to filtered tap-water and standard pelleted laboratory chow (U.A.R., France) throughout the studies. Carrageenan lambda type IV (Sigma, France) was prepared as a 2% (w/v) solution in saline. Compounds to be tested were dissolved in dimethylsulfoxide (DMSO) such that the indicated doses were in a final volume of 0.05 mL. Doses were expressed as mg/paw free active substance. From 17 to 19 hours before the studies the rats were fasted with free access to water. The paw volumes of the

left hindpaws of the rats were measured using an electronic plethysmometer type 7140 (Ugo Basile – Italy) at time = 0. Paw edema was then induced by injection of 0.05 mL of 2% carrageenan solution into the left hindpaws of the rats. Immediately after injection of the carrageenan, compounds in DMSO or vehicle alone were injected into the same paw in a volume of 0.05 mL in a blind and random fashion.

[0047] The paw volumes were measured at 1.5 hours, 3 hours, 4.5 hours and 24 hours after administration of the compounds. The edema volume of each rat at each time point was expressed as the change from the initial paw volume (time = 0). A total of 5 rats were used for each compound dose and the average edema volume calculated for each dose. The anti-inflammatory effect in treated groups was expressed as the percent inhibition of edema volume compared to the vehicle-treated group at 1.5 hours, 3 hours, 4.5 hours and 24 hours.

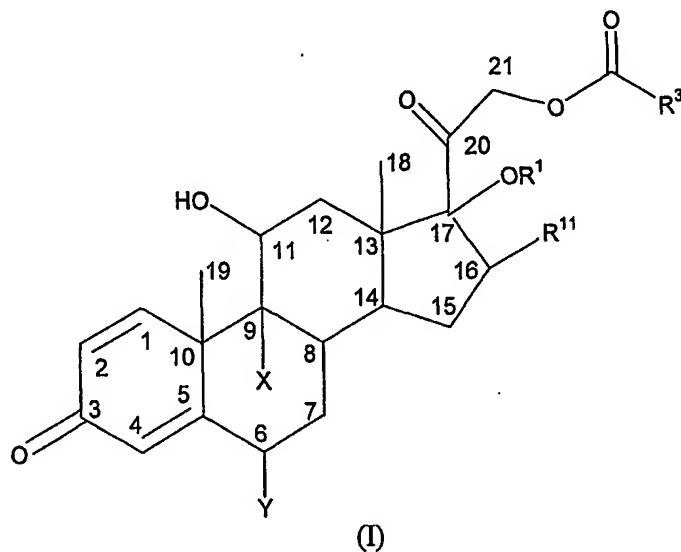
[0048] The results are shown in Figure 1, in which the efficacies of equimolar doses (equivalent to 10 µg per paw of budesonide) are compared at 24 hours. Budesonide itself reduces swelling by 46%. The formation of a carbonate ester at C21 decreases the efficacy of budesonide when the ester is C<sub>6</sub> or smaller. Unexpectedly, at C<sub>6</sub> the curve reverses, and the efficacy increases. Thus, although one would expect C<sub>7</sub> to be less efficacious than C<sub>6</sub>, in fact it is surprisingly found more efficacious, and the C<sub>10</sub> carbonate is 30% more efficacious than budesonide itself. The correlation between numbers of carbons in the carbonate ester and efficacy reaches a peak at C<sub>12</sub> with an 88% reduction of swelling.

[0049] Other carbonates showed similar behavior. The phytol (Example 16) and farnesol (Example 17) carbonates exhibited normal onset of action and maximum activity at 24 hours of 69 and 86 percent respectively. The amine-terminal alkylcarbonate, Example 15, exhibited a maximum activity at 24 hours of 73 percent.

[0050] The enhanced effect of the carbonates and carbamates compared to the parent steroid is most dramatic at the 24-hour observation, as can be seen by comparing Figures 2 and 3, in which budesonide carbonate is compared to budesonide.

## CLAIMS

1. A compound of Formula I:



wherein

$R^1$  and  $R^2$ , independently for each occurrence, represent a hydrogen, lower alkyl or lower acyl, or taken together  $R^1$  and  $R^2$  form a substituted or unsubstituted ketal;

$R^3$  is  $-OR^4$  or  $-NR^5R^6$ ;

$R^4$  is chosen from  $C_7$  to  $C_{24}$  hydrocarbon,  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-COOH$  and  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-NR^9R^{10}$ ;

$R^5$  is hydrogen or  $C_7$  to  $C_{24}$  hydrocarbon;

$R^6$  is chosen from  $C_7$  to  $C_{24}$  hydrocarbon and  $-(C_7 \text{ to } C_{24} \text{ hydrocarbon})-COOH$ ;

$R^9$  is hydrogen or  $C_1$  to  $C_{17}$  hydrocarbon;

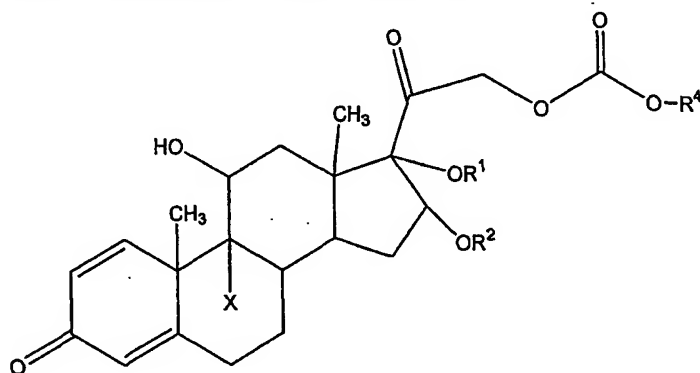
$R^{10}$  is hydrogen or  $C_1$  to  $C_{17}$  hydrocarbon;

$R^{11}$  is methyl or  $-OR^2$ ; and

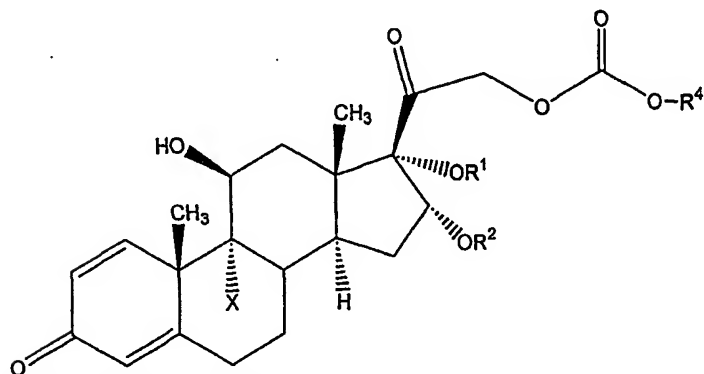
X and Y are independently hydrogen or halogen.



2. A compound according to claim 1 of formula:

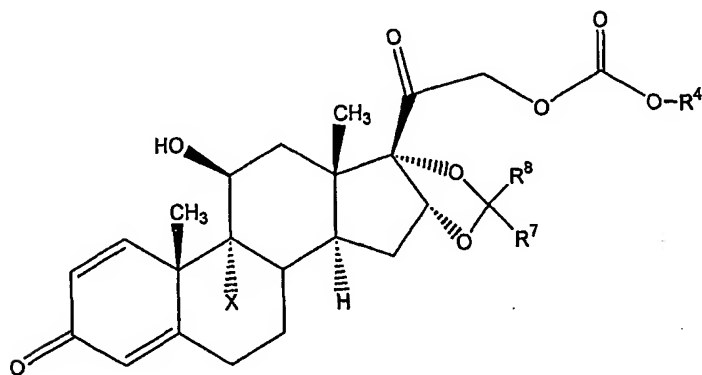


3. A compound according to claim 2 of formula:



wherein X is hydrogen or fluorine.

4. A compound according to claim 3 of formula

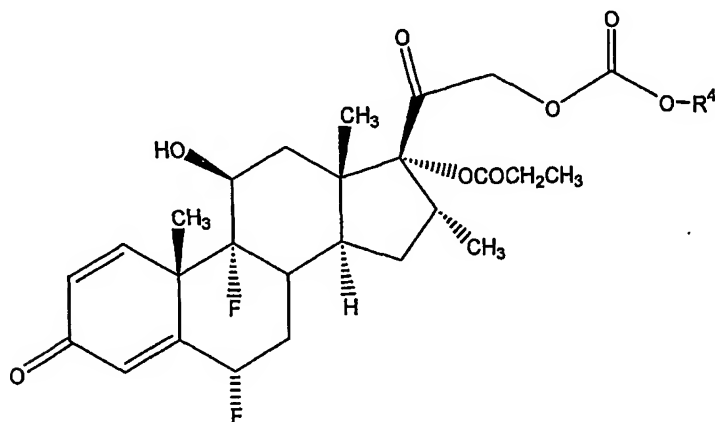


wherein

$R^7$  is hydrogen or lower alkyl; and

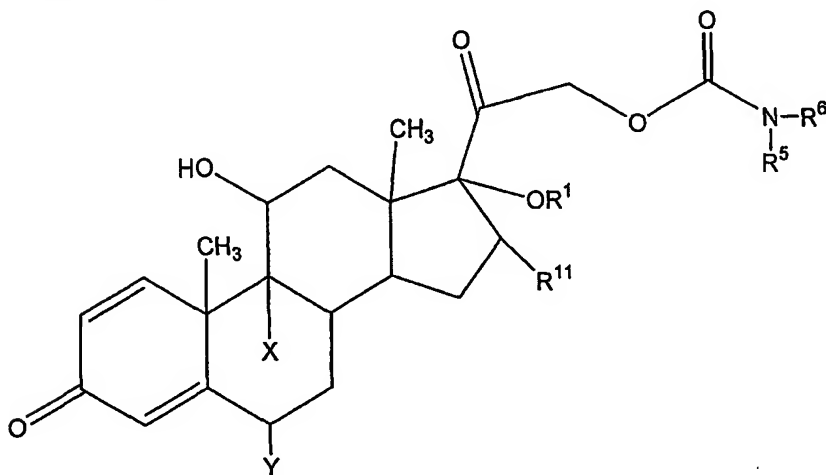
$R^8$  is lower alkyl.

5. A compound according to claim 1 of formula

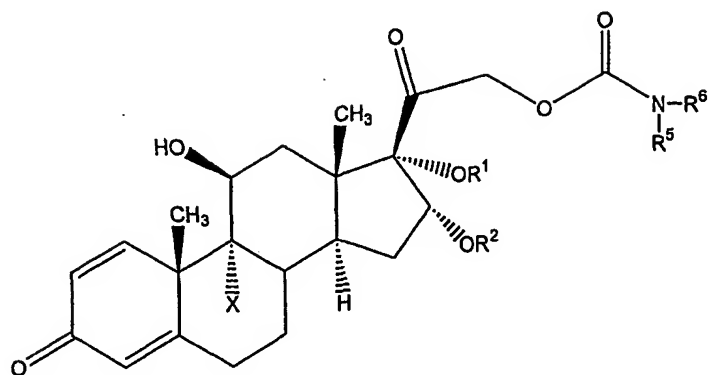


6. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{11}$  to  $C_{14}$  alkyl.
7. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{11}$  to  $C_{18}$  alkyl.
8. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{12}$  to  $C_{24}$  alkyl.
9. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{12}$  to  $C_{20}$  alkyl.
10. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_7$  to  $C_{24}$  alkyl.
11. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_8$  to  $C_{24}$  alkyl.
12. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_9$  to  $C_{24}$  alkyl.

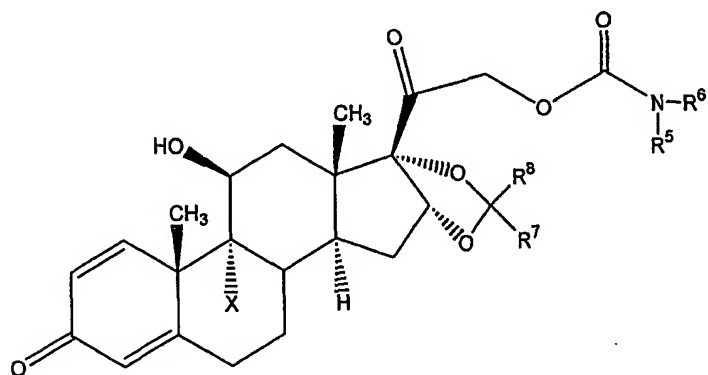
13. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{10}$  to  $C_{24}$  alkyl.
14. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{11}$  to  $C_{24}$  alkyl.
15. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_8$  to  $C_{18}$  alkyl.
16. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_{10}$  to  $C_{16}$  alkyl.
17. A compound according to any of claims 1 to 5 wherein  $R^4$  is  $C_8$  to  $C_{20}$  alkyl.
18. A compound according to claim 1 of formula:



19. A compound according to claim 18 of formula:



20. A compound according to claim 19 of formula

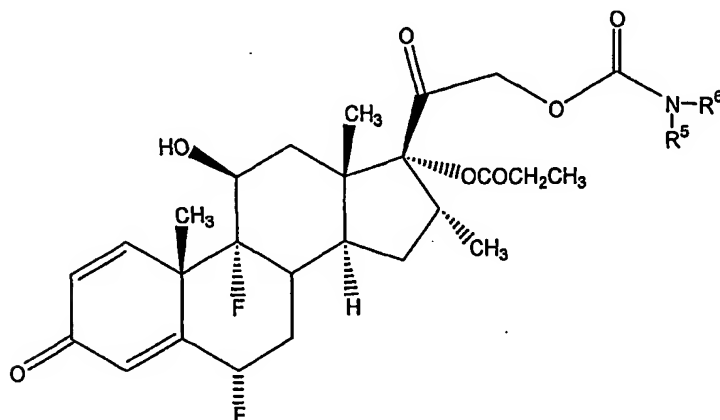


wherein

R<sup>7</sup> is hydrogen or lower alkyl; and

R<sup>8</sup> is lower alkyl.

21. A compound according to claim 18 of formula



22. A compound according to any of claims 18 to 21 wherein R<sup>5</sup> is hydrogen or lower alkyl.
23. A compound according to any of claims 18 to 21 wherein R<sup>6</sup> is C<sub>11</sub> to C<sub>14</sub> alkyl.
24. A compound according to any of claims 18 to 21 wherein R<sup>6</sup> is C<sub>11</sub> to C<sub>18</sub> alkyl.
25. A compound according to any of claims 18 to 21 wherein R<sup>6</sup> is C<sub>12</sub> to C<sub>24</sub> alkyl.
26. A compound according to any of claims 18 to 21 wherein R<sup>6</sup> is C<sub>12</sub> to C<sub>20</sub> alkyl.
27. A compound according to any of claims 18 to 21 wherein R<sup>6</sup> is C<sub>7</sub> to C<sub>24</sub> alkyl.
28. A compound according to any of claims 18 to 21 wherein R<sup>6</sup> is C<sub>8</sub> to C<sub>24</sub> alkyl.



36. A method for treating rhinitis comprising administering the compound of any of claims 1-35.
37. A method for treating asthma comprising administering the compound of any of claims 1-35.
38. A method according to either of claims 36 or 37 wherein said compound is administered by inhalation.
39. A method for treating a disorder chosen from osteoarthritis, bursitis, epicondylitis, tenosynovitis, lichen simplex chronicus, granuloma annulare, lichen planus, keloids, alopecia areata, discoid lupus erythematosus, localised neurodermatitis, cystic acne, granuloma annulare, nummular and dyshydrotic eczema, hypertrophic scars and macular degeneration comprising administering the compound of any of claims 1-35.
40. A pharmaceutical formulation for inhalation comprising a compound according to any of claims 1-35 and a pharmaceutically acceptable fluid for suspension or solution.
41. A pharmaceutical formulation for inhalation according to claim 40 additionally comprising a propellant.
42. A pharmaceutical formulation for topical application comprising a compound according to any of claims 1-35 and a pharmaceutically acceptable carrier for topical or transdermal application.

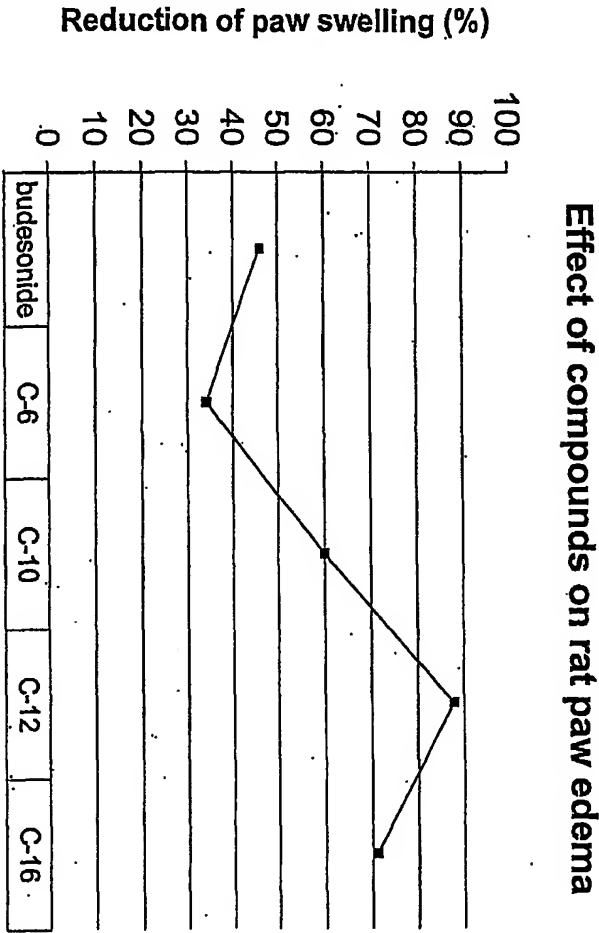
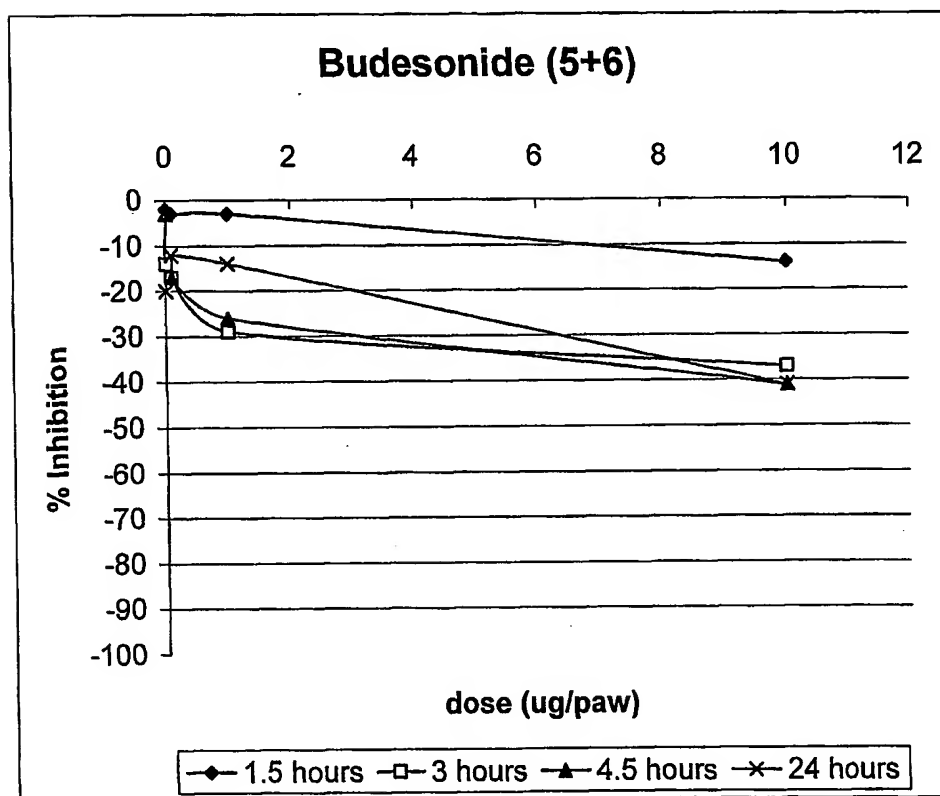
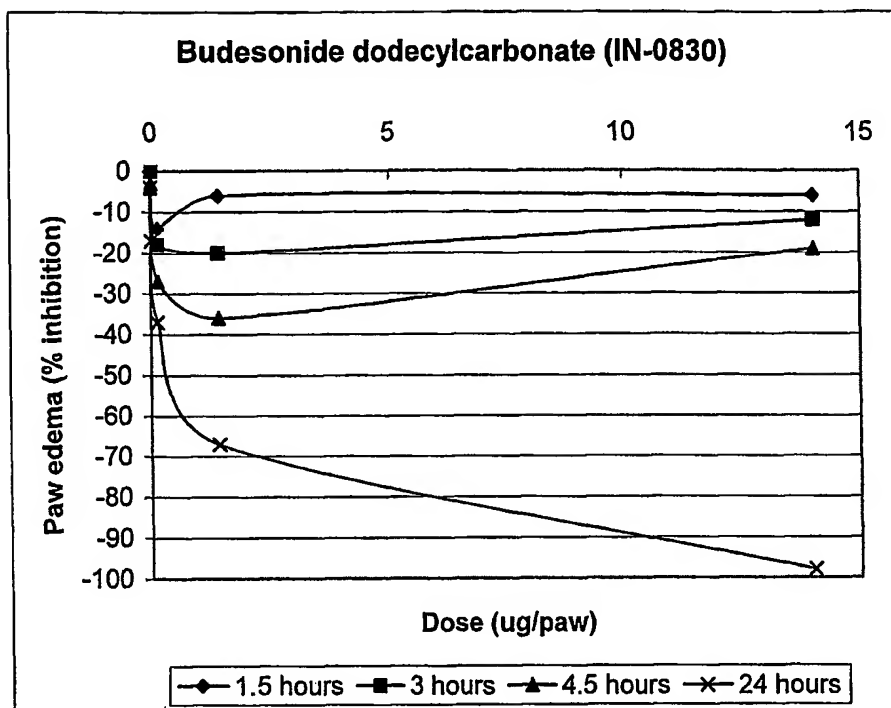


FIG. 1



**FIG. 2**

**FIG. 3**